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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.013 Å R factor = 0.054 wR factor = 0.185 Data-to-parameter ratio = 8.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2H-Chromene-2-thione

The crystal structure of the title compound, C_9H_6OS , also known as 2-thiocoumarin, has been determined. The molecule is essentially planar and crystals may display second harmonic generation (SHG) effects.

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Comment

Coumarin has been identified as an important compound in the area of photochemistry, especially as a building block in the crystal lattice for photo-cycloaddition reactions (Vishnumurthy et al., 2001). Coumarin derivatives are known to generate quality dyes, and are prominent in natural products chemistry (Dean, 1963; Murray et al., 1982). They have been found to possess a wide variety of uses in the perfume industry, as flavour enhancers, sunscreens, laser dyes (Khalfan et al., 1987), and in the pharmaceutical industry (Hooper et al., 1982; Morris & Russell, 1971). Theoretical calculations using semiempirical methods demonstrate that coumarin and its sulfur derivatives possess well defined residual dipole moments and the title compound, 2H-chromene-2-thione, (II), crystallizes in a non-centrosymmetric space group. These features make coumarin and its derivatives excellent candidates for organic SHG materials. We have synthesized a series of coumarins in this context (see Scheme below). Here we report the structure of (II). The molecular structure of (II) is illustrated in Fig. 1. Fig. 2 shows the packing of the molecules in the crystal lattice. Since the data were restricted to a unique octant of reflections only, the Flack (1983) parameter does not give a reliable indication of the absolute configuration.



Experimental

The title compound was synthesized in a single step (Scheibye *et al.*, 1979). Compound (I) was converted to compound (II) with an equimolar amount of methoxyphenylthionophosphine sulfide (Lawesson's Reagent). Crystals were grown by slow evaporation at low temperature from ethanol as yellow needles (m.p. 371 K).

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Figure 1

The molecular structure of (II), showing the atomic numbering. Displacement ellipsoids are drawn at the 50% probability level. H atoms are shown as spheres of arbitrary radii.

Crystal data

 $\begin{array}{l} C_9H_6OS\\ M_r = 162.21\\ Orthorhombic, P2_12_12_1\\ a = 4.126 \ (1) \ \text{\AA}\\ b = 17.779 \ (3) \ \text{\AA}\\ c = 10.268 \ (3) \ \text{\AA}\\ V = 753.2 \ (3) \ \text{\AA}^3\\ Z = 4\\ D_x = 1.431 \ \text{Mg m}^{-3} \end{array}$

Data collection

Enraf–Nonius Turbo-CAD-4 diffractometer Non-profiled $\omega/2\theta$ scans Absorption correction: none 815 measured reflections 815 independent reflections 375 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.054$ $wR(F^2) = 0.186$ S = 0.84815 reflections 100 parameters H-atom parameters constrained Mo K α radiation Cell parameters from 25 reflections $\theta = 6.1-13.3^{\circ}$ $\mu = 0.36 \text{ mm}^{-1}$ T = 293 (2) K Needle, yellow $0.38 \times 0.05 \times 0.04 \text{ mm}$

D., not measured

 $\theta_{\max} = 25.0^{\circ}$ $h = 0 \rightarrow 4$ $k = 0 \rightarrow 21$ $l = 0 \rightarrow 12$ 3 standard reflections frequency: 60 min intensity decay: 3%

 $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.1169P)^{2}]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.20 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{min} = -0.26 \text{ e} \text{ Å}^{-3}$ Absolute structure: Flack (1983) Flack parameter = 0.1 (5)

Data collection: CAD-4 EXPRESS (Enraf–Nonius, 1994); cell refinement: CAD-4 EXPRESS; data reduction: XCAD4 (Harms &



Figure 2

Packing diagram of the title compound, viewed down the *a* axis.

Wocadlo, 1995); program(s) used to solve structure: *SIR*92 (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *ORTEP*3 for Windows (Farrugia, 1997); software used to prepare material for publication: *PLATON* (Spek, 1990).

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